## ORIGINAL PAPER

# An efficient and environmentally friendly procedure for synthesis of pyrimidinone derivatives by use of a Biginelli-type reaction

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**Abstract** A Biginelli-type three-component reaction involving cyclopentanone, aromatic aldehyde, and urea or thiourea for preparation of pyrimidinone derivatives under neat conditions is described. This condensation reaction can also take place smoothly in the presence of vitamin  $B_1$ in EtOH at 80 °C in good yield. The procedure is simple, high-yielding, time-saving, and environment friendly.

**Keywords** Multicomponent reactions · Aldehyde · Organocatalyst · Green chemistry

### Introduction

Multicomponent reactions (MCRs), by virtue of their convergence, productivity, facile execution, and generally high yield of products, have attracted much attention [1-4]. The century-old Biginelli reaction as a classical multi-component reaction has gained much importance in organic synthesis, partly because of the diverse types of physiological activity associated with the dihydropyrimid-inones produced by this reaction [5-7].

The classical Biginelli reaction is a simple one-pot cyclocondensation of  $\beta$ -dicarbonyl compounds with aldehydes and urea or thiourea in the presence of various

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Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai 201203, People's Republic of China e-mail: simmhulh@mail.shcnc.ac.cn catalysts [8–15]. Pan et al. [16] reported an efficient alternative for synthesis of fused pyrimidinones by a three-component condensation with aromatic aldehyde, cyclopentanone, and urea or thiourea. However, the use of stoichiometric amounts of TMSCl as additional reagent and mixed DMF/CH<sub>3</sub>CN as reaction solvent seemed to be necessary to obtain satisfactory results. Furthermore, TMSCl is toxic, corrosive, and environmental unfriendly. More recently, Zhang described the three-component condensation of aromatic aldehyde, cyclopentanone, and urea or thiourea in the presence of YbCl<sub>3</sub> [17]. For conservation of the environment combined with economic aspects, the application of metal ion-free, environmentally safe, and convenient reagents in synthetic and medicinal chemistry is in demand [18–20].

Thus, the development of a simple, efficient, and environmentally friendly method for the Biginelli-type reaction is needed. Herein, we wish to report a novel and efficient procedure for synthesis of pyrimidinone derivatives via the three-component condensation of aromatic aldehyde, cyclopentanone, and urea or thiourea under neat conditions (Scheme 1) or in the presence of vitamin  $B_1$  as an organocatalyst in EtOH (Scheme 2).

### **Results and discussion**

Initially, we studied the Biginelli-type reaction of benzaldehyde (1a, 5 mmol), cyclopentanone (2, 5 mmol), and urea (3a, 6 mmol) under solvent-free conditions at 140 °C. To our delight, reaction was complete in 10 min, giving the desired product 4a in 92% yield. Encouraged by the result, a series of aldehydes were selected to undergo the condensation under solvent-free conditions at 140 °C (Table 1).

#### Scheme 1

Scheme 2



Table 1 Synthesis of Biginelli-type product 4

Entry	Ar	X	Time (min)	Product	Yield (%) <sup>a</sup>	Ref.
1	C <sub>6</sub> H <sub>5</sub> (1a)	0	10	4a	92	[16]
2	$4-MeC_{6}H_{4}$ (1b)	0	15	4b	95	[ <mark>16</mark> ]
3	$4-MeOC_{6}H_{4}$ (1c)	0	25	4c	85	[ <mark>16</mark> ]
4	$3-NO_2C_6H_4$ (1d)	0	15	4d	86	[ <mark>16</mark> ]
5	$4-NO_2C_6H_4$ (1e)	0	15	<b>4e</b>	86	[ <b>17</b> ]
6	$4-ClC_{6}H_{4}$ (1f)	0	10	4f	90	[ <mark>16</mark> ]
7	4-FC <sub>6</sub> H <sub>4</sub> ( <b>1g</b> )	0	15	4g	90	[ <mark>16</mark> ]
8	$4\text{-CNC}_6\text{H}_4$ (1h)	0	15	4h	88	[ <b>17</b> ]
9	3-MeO-4-OHC <sub>6</sub> H <sub>3</sub> (1i)	0	20	<b>4i</b>	_ <sup>b</sup>	
10	2,4-Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub> ( <b>1j</b> )	0	20	4j	20	
11	$4\text{-}BzOC_{6}H_{4} (\mathbf{1k})$	0	20	4k	25	
12	$C_{6}H_{5}$ (1a)	S	60	41	82	[ <mark>16</mark> ]
13	$4-MeC_{6}H_{4}$ (1b)	S	60	4m	84	[ <b>17</b> ]
14	$4-MeOC_{6}H_{4}$ (1c)	S	60	4n	80	[ <mark>16</mark> ]
15	$4-NO_2C_6H_4$ (1d)	S	60	40	82	[16]
16	$4-ClC_{6}H_{4}$ (1f)	S	60	4p	86	[16]

Conditions: aromatic aldehyde (1, 5 mmol), cyclopentanone (2, 5 mmol), and urea (3a) or thiourea (3b) (6 mmol), 140  $^{\circ}$ C

<sup>a</sup> Isolated yield

<sup>b</sup> Several side reactions were observed and the desired product could not be isolated

As shown in Table 1, aromatic aldehydes bearing functional groups (for example –H, –CH<sub>3</sub>, –OCH<sub>3</sub>, –Cl, –F, –NO<sub>2</sub>, and –CN) react smoothly to give the corresponding products in good yields (80–95%). We also found that this condensation reaction was complete within 60 min, much quicker than reported methods (3–10 h) [16, 17].

However, several side reactions were observed when aromatic aldehydes bearing the –OH were used as a starting material and the desired product could not be isolated (Table 1, entry 9). In addition, very low yields of the target products were obtained when using 1j and 1k as substrates because the reaction mixture could not be efficiently stirred (Table 1, entries 10 and 11). Hence, we turned our attention towards the synthesis of pyrimidinone derivatives 4i–4k by condensation of cyclopentanone (2), urea (3a), and aromatic aldehydes 1i–1k.

As shown in Table 2, the mixture of 4-hydroxy-3-methoxybenzaldehyde (1i), cyclopentanone (2), and urea (3a) was chosen as the model reaction under different reaction conditions. First, we changed the reaction temperature from 20 to 140 °C (Table 2, entries 1–4), because, unfortunately, no desired product 4i was observed when the mixture was stirred at 20 °C for 10 h (Table 2, entry 1). In addition, several side reactions were observed, as indicated by TLC, when the three-component condensation reaction took place at 80–140 °C under solvent-free conditions (Table 2, entries 2–5). We also carried out the condensation reaction at 80 °C in various solvents (Table 2, entries 5–10), for example MeOH, EtOH, THF, DMF, CH<sub>3</sub>CN, and DMSO. Unfortunately, only trace amounts of the target product were observed when the mixture was stirred at 80 °C in these solvents.

In our previous study, we found that several multi-component reactions proceeded smoothly in the presence of vitamin  $B_1$  as catalyst [21]. Therefore, we carried out the reaction in EtOH at 80 °C, taking a 1:1:1.2 mol ratio of 4-hydroxy-3-methoxybenzaldehyde (1i), cyclopentanone (2), and urea (3a) in the presence of 3 mol% vitamin  $B_1$  for 3 h. To our delight, the desired product 4i was obtained in 86% yield (Scheme 2). We changed the amount of vitamin  $B_1$  from 1 to 10 mol%, and found use of 3 mol% was optimum to ensure

 Table 2 Synthesis of Biginelli-type product 4i under various conditions

Entry	Solvent	Catalyst (mol%)	Temp. (°C)	Time (h)	Yield (%)
1	None	0	20	10	NR <sup>b</sup>
2	None	0	80	1	_ <sup>c</sup>
3	None	0	120	0.3	_ <sup>c</sup>
4	None	0	140	0.3	_ <sup>c</sup>
5	MeOH	0	80	3	Trace
6	EtOH	0	80	3	Trace
7	THF	0	80	3	Trace
8	DMF	0	80	3	Trace
9	CH <sub>3</sub> CN	0	80	3	Trace
10	DMSO	0	80	3	Trace
11	EtOH	Vitamin $B_1(1)$	80	3	45
12	EtOH	Vitamin $B_1(3)$	80	3	86
13	EtOH	Vitamin $B_1(5)$	80	3	86
14	EtOH	Vitamin $B_1$ (10)	80	3	85

Conditions: 4-hydroxy-3-methoxybenzaldehyde (1i, 5 mmol), cyclopentanone (2, 5 mmol), and urea (3a, 6 mmol), solvent-free or solvent  $3 \text{ cm}^3$ 

<sup>a</sup> Isolated yield

<sup>b</sup> No reactions were observed

 $^{\rm c}\,$  Several side reactions were observed and the desired product could not be isolated

high reaction efficiency (Table 2, entries 11–14). Then, we carried out the three-component condensation reaction using aromatic aldehydes 1i-1k in the presence of 3 mol% vitamin B<sub>1</sub> in EtOH for 3 h and obtained the desired products 4i-4k in 86, 83, and 90% yield, respectively (Scheme 2).

In conclusion, this method is a new and simple modification of the Biginelli-type reaction of aromatic aldehyde, cyclopentanone, and urea or thiourea. The three-component condensation reaction can proceed under solvent-free conditions at 140 °C without any catalysts in excellent yields (80-95%). In addition, the condensation reaction can also take place smoothly in the presence of 3 mol% vitamin  $B_1$  in EtOH at 80 °C in good yields (83-90%). It is important to note that the reaction can proceed smoothly in good yield by employing a catalytic amount of vitamin B<sub>1</sub> as the catalyst when using an aldehyde carrying a hydroxy group as the starting material, which normally gives an extremely poor yield under catalyst-free conditions. This one-pot synthesis of hydropyrimidinone derivatives is therefore simple, high yielding, time saving, and environmentally friendly. It is an useful addition to existing methods.

## Experimental

Reagents and all solvents were analytically pure grade and were used without further purification.  ${}^{1}$ H and  ${}^{13}$ C NMR

spectra were recorded on a Varian 400 MHz spectrometer at 400 and 100 MHz, with TMS as an internal standard. Chemical shifts ( $\delta$ ) are given in ppm relative to TMS, coupling constants (*J*) in Hz. Melting points were determined with an X-4 apparatus and are corrected. Mass spectra were measured with a Thermo Finnigan LCQ-Advantage. Elemental analyses (C, H, N, S) were performed on a Fisons EA 1110 CHNS elemental analyzer; results agreed favorably with calculated values.

# General procedure for synthesis of compounds 4a–4h and 4l–4p

An aldehyde (1a–1h, 5 mmol), cyclopentanone (2, 5 mmol), and urea (3a) or thiourea (3b) (3.6 mmol) were mixed in a round-bottomed flask. The mixture was stirred at 140 °C for the time needed to complete the reaction (monitored by TLC). After cooling to room temperature, the reaction was quenched with 20 cm<sup>3</sup> H<sub>2</sub>O and stirred for 10 min. The pure product 4 was isolated by filtration, followed by washing with EtOAc.

# General procedure for the synthesis of compounds 4i-4k

A mixture of aldehyde (**1i–1k**, 5 mmol), cyclopentanone (**2**, 5 mmol), urea (**3a**, 3.6 mmol), and vitamin B<sub>1</sub> (0.15 mmol) in 3 cm<sup>3</sup> ethanol was heated to 80 °C, with stirring, for 3 h to complete the reaction (monitored by TLC). After cooling to room temperature, the reaction was quenched with 20 cm<sup>3</sup> H<sub>2</sub>O and stirred for 10 min. The pure product was isolated by filtration, followed by washing with EtOAc.

# 1,3,4,5,6,7-Hexahydro-7-(4-hydroxy-3-methoxybenzylidene)-4-(4-hydroxy-3-methoxyphenyl)-2H-cyclopenta[d]pyrimidin-2-one (**4i**, C<sub>22</sub>H<sub>22</sub>N<sub>2</sub>O<sub>5</sub>)

Yellow solid, yield 86%; m.p.: 241–244 °C; <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ):  $\delta = 9.03$  (brs, 1H), 8.93 (brs, 1H), 8.60 (brs, 1H), 7.04 (s, 1H), 6.88 (s, 1H), 6.81 (d, J = 2 Hz, 1H), 6.75–6.78 (m, 3H), 6.67 (dd,  $J_1 = 2.0$  Hz,  $J_2 = 8.4$  Hz, 1H), 6.52 (brs, 1H), 5.04 (brs, 1H), 3.78 (s, 3H), 3.76 (s, 3H), 2.77–2.85 (m, 2H), 2.32–2.41 (m, 1H), 2.02–2.11 (m, 1H) ppm; <sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ ):  $\delta = 153.7$ , 148.0, 147.9, 146.4, 145.7, 136.5, 136.3, 134.9, 129.9, 121.6, 119.4, 117.6, 117.2, 116.1, 115.9, 112.3, 111.3, 57.6, 56.1, 55.9, 28.9, 28.7 ppm; MS (ESI): m/z =417 ([M + Na]<sup>+</sup>).

# 7-(2,4-Dichlorobenzylidene)-4-(2,4-dichlorophenyl)-

 $1,3,4,5,6,7\mbox{-}hexahydro-2H\mbox{-}cyclopenta[d]pyrimidin-2\mbox{-}one (4j, C_{20}H_{14}Cl_4N_2O)$ 

White solid, yield 83%; m.p.: 251-254 °C; <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ):  $\delta = 9.19$  (brs, 1H), 7.33–7.68 (m, 6H), 7.25 (s, 1H), 6.74 (brs, 1H), 5.6 (s, 1H), 2.62–2.83

(m, 2H), 2.37–2.46 (m, 1H), 1.91–2.04 (m, 1H) ppm; <sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ ):  $\delta = 153.6$ , 142.6, 139.8, 136.9, 134.9, 133.7, 133.3, 132.3, 131.7, 131.1, 130.6, 129.3, 129.2, 128.8, 128.4, 127.6, 113.0, 54.5, 28.4, 28.3 ppm; MS (EI): m/z = 442, 440, 438 (M<sup>+</sup>).

# 7-[4-(Benzyloxy)benzylidene]-4-[4-(benzyloxy)phenyl]-1,3,4,5,6,7-hexahydro-2H-cyclopenta[d]pyrimidin-2-one (**4k**, C<sub>34</sub>H<sub>30</sub>N<sub>2</sub>O<sub>3</sub>)

Yellow solid, yield 90%; m.p.: 201–204 °C; <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ):  $\delta = 8.68$  (brs, 1H), 6.95–7.45 (m, 19H), 6.56 (s, 1H), 5.03–5.13 (m, 5H), 2.67–2.78 (m, 2H), 1.95–2.04 (m, 1H), 1.78–1.90 (m, 1H) ppm; <sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ ):  $\delta = 165.2$ , 158.2, 153.7, 143.1, 143.0, 137.6, 137.4, 136.2, 136.1, 131.1, 131.0, 129.6, 128.9, 128.8, 128.3, 128.2, 128.1, 118.1, 116.6, 115.4, 115.3, 69.7, 69.6, 57.3, 28.7, 28.8 ppm; MS (ESI): m/z = 537([M + Na]<sup>+</sup>).

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